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Chemical Reactions (page 49) 1. What is a
chemical reaction? It is a process that
changes one set of chemicals into another

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set of chemicals. 2. Complete the table about chemicals in a chemical reaction.

Energy-Releasing Reaction Energy Course
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Reactants The elements or compounds that enter into a chemical reaction Products

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Chemical reactions. The Role of Electrons (pages 25–28) 13. The space in which the electrons move is huge compared to the space occupied by the . 14. The electrons farthest from the nucleus or most loosely held are called . 15.

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For most chemical reactions in this book, solids are labeled (s), liquids (l), and gases (g). The numerical coefficients in front of the chemical formulas express the moles of each compound or element. The preceding reaction can be interpreted in terms of moles or masses. (See Table 1.)

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