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27 A 2+ B 3+ C 2– D 3– 4. Fe. 2. O. 3 + x. CO. y. Fe + 3CO. 2. This equation will be balanced when. A . x = 1 and . y = 2 B . x = 2 and . y = 2 C . x = 3 and . y = 2 D . x = 2 and . y = 3. 5. An acidic solution contains A only hydrogen ions B only hydroxide ions C more hydrogen ions than hydroxide ions D more hydroxide ions than hydrogen ions. 6.

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2 (g) → 3CO 2 (g) + 4H 2O(l) Name the type of enthalpy change which the student measured. No marks should be awarded for 'burning' since the word "burned" appears in the text. (o) Unless the question is clearly about a non-chemistry issue, eg costs in industrial chemical process, a non-chemical answer gains no marks.

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2 by electricity / (electric) current / (flow of) electrons Mark independently b A = chlorine / Cl2 Ignore Cl 3 B = hydrogen / H2 Ignore H C = sodium hydroxide / NaOH Ignore references to sodium chloride If both name and formula given, both must be correct, but ignore Cl and H Award 1 mark for chlorine and hydrogen the wrong way round

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